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Paramagnetic Resonance Spectra of Rare Earth Ions in the Crystal Field of Calcium Fluoride*

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We have studied the paramagnetic resonance spectra of rare earth ions in the crystal field of calcium fluoride. These spectra can be classified as belonging to three types: a. spectra which indicate that the rare earth ions are exposed to a cubic field; b. spectra in which the anisotropic behaviour suggests an axial crystal field along the cubic directions; c. spectra in which the axial distortion is along the body diagonals of the cube.

The calcium ion with fluorite structure is surrounded by a cube of eight nearest fluorine ions. The trivalent rare earth ions substitute for the calcium ion. However, a charge compensation has to take place to preserve the neutrality of the crystal. From measurements of the x-ray lattice constants¹⁾ and of the ionic conductivity²⁾ it has been shown that in well annealed crystals the main disorder introduced into the crystal by the addition of the rare earth ion is the formation of F⁻ interstitial (see Fig. 1). The empty sites which are available for the inter-





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stitial F^- ion are located at the center of the adjacent cube along one of the cubic axes. Supporting evidence for the F^- interstitial formation is found in the work of Bleaney *et al.*³⁾.

With proper thermal treatment, part or a large fraction of the interstitial fluorine ions can be removed from the adjacent cube next to the rare earth ion4). In this case, the point symmetry of the rare earth ions approximates O_h . In some crystals, however, the predominant axial distortion is found from paramagnetic resonance spectra to be along the body diagonals of the cube. The optical spectra shows a number of strong bands and very poor ultraviolet transmission. It is likely that one of the fluorine ions at the corner of the cube is replaced by O²⁻ or OH⁻ ion. Supporting evidence is found that on heating the crystal in air this spectrum increases in intensity.

a)
$$V_{\text{cubic}} = B_4{}^0[O_4{}^0 + 5(O_4{}^4 + O_4{}^{-4})] + B_6{}^0[O_6{}^0 - 21(O_6{}^4 + O_6{}^{-4})] \quad (1)$$

b)
$$V_{\text{axial}} = B_2{}^0O_2{}^0 + B_4{}^0O_4{}^0 + B_6{}^0O_6{}^0 + B_4{}^4(O_4{}^4 + O_4{}^{-4}) + B_6{}^4(O_6{}^4 + O_6{}^{-4})$$

c)
$$V_{\text{diagonal}} = B_2{}^0O_2{}^0 + B_4{}^0O_4{}^0 + B_6{}^0O_6{}^0 + B_4{}^3(O_4{}^3 + O_4{}^{-3}) + B_6{}^3(O_6{}^3 + O_6{}^{-3})$$

where B_n^m are crystal field coefficients proportional to r^n/R^{n+1} and O_n^m are operators transforming the corresponding spherical Legendre polynomials. The magnetic properties of these three types of spectra are very different and we shall illustrate this in a few selected examples. A full report of the magnetic properties of the rare earth ions in calcium fluoride will appear elsewhere.

Er³⁺:
$${}^{4}H_{15/2}$$

a) In the cubic field of calcium fluoride, the 16-fold degeneracy splits into 3 quartets (Γ_8) and two doublets. The lowest levels are given by

$$-258.86 c - 162 d, \Gamma_8$$

 $-26 c - 312 d, \Gamma_7$

where c and d are parameters signifying the fourth and sixth order contribution. For a point charge model they are given by

$$c = \frac{2^3}{3^3.11.13} \frac{ze^2}{R^5} \quad d = \frac{2^4}{3^3} \frac{.10}{11.13^2} \frac{ze^2}{R^7}$$

The observed g factor is 6.785 ± 0.002^{5} . This is consistent with a Γ_{τ} ground state which gives $g_{\text{calculated}} = 6.79$. A lowest Γ_{τ} level indicates that the rare earth ion spectrum is dominated to a large extent by the sixth order term in the potential.

b) An axial spectrum has been observed by Baker *et al*⁶, and by the author with g_{\parallel} =7.76 and g_{\perp} =6.253. This is consistent with the same doublet level except now exposed to an axial field since $(g_{\parallel}+2g_{\perp})/3\sim g_{\text{euble.}}$

c) A set of three lines, arising from the three inequivalent sites with axial symmetry along the three body diagonal directions, is observed in some crystals⁷. Each ion can be discribed by g factors $g_{\parallel}=2.23$, $g_{\perp}=8.9$, and again $(g_{\parallel}+2g_{\perp})/3 \sim g_{\text{cubic.}}$ The g factors can be explained to arise from a doublet with

 $a|\pm 13/2 > +b|\pm 7/2 > +c|\pm 1/2 > +d|\mp 5/2 > +e|\mp 11/2 >$

with the coefficients b and d larger than a, c or e.

(a) The cubic field suggests that Γ_7 is the

ground state level. The g wave function of the doublet is given by

$$\sqrt{3/4} |\pm 5/2 > -\sqrt{1/4} |\mp 3/2 >$$

and the calculated g factor = 24/7. The observed g factor of 3.441 ± 0.001 is in essential agreement⁸⁾.

(c) The spectrum along the body diagonals can be represented by the spin Hamiltonian

$$\mathcal{H} = g_{\parallel}\beta H_z S_z + g_{\perp}\beta (H_x S_x + H_y S_y) + AS_z I_z + B(S_x I_x + S_y I_y) + P(I_z^2 - \frac{1}{3}I(I+1)) .$$

with
$$S=\frac{1}{2}$$
 and $I=\frac{1}{2}$ or $I=5/2$

The observed g factor is $g_{\parallel} = 1.323 \pm 0.001$ and $g_{\perp} = 4.387 \pm 0.002$ showing that this is the Γ_{τ} anisotropic level. This Kramers doublet can be approximated with

$$0.940|\pm\frac{1}{2}>+0.264|\pm7/2>+0.209|\mp5/2>$$

The hyperfine part is expressed as

$$\begin{array}{cccc} A^{171} = 354.6 & B^{171} = 1168 & P = 85 \pm 3 \\ A^{173} = 96 & B^{173} = 320 & (g_{\parallel} B/g_{\perp} A) \sim 1 \end{array}$$

all in 10⁻⁴ cm⁻¹. These results indicate a large quadrupole moment of about 3.1 barns.

Nd3+: 4/9/2.

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(a) Calculations show that the ground state for all reasonable ratios of the fourth to sixth order parameters is given by a quartet Γ_8 . The wave functions and g factors are given by

g value	transition and been shall be transition	intensity
2.24	$0.8722 -5/2 > -0.4892 +3/2 > \rightarrow 0.8722 +5/2 > -0.4892 -3/2 > -0.4$	20
1.15	$0.8722 \pm 5/2 > -0.4892 \mp 3/2 > \rightarrow 0.05410 \mp 9/2 > -0.2981 \pm 1/2 > +0.9530 \pm 7/2 > -0.2981 \pm 7/2 > -0.2981 \pm 7/2 > +0.9530 \pm 7/2 = -0.9530 \pm 7/2 = -0.9570 \pm -0.9570 \pm 7/2 = -$	4.4
4.54	$0.5410 +9/2>-0.2981 \pm 1/2>+0.9530 -7/2>\rightarrow 0.05410 -9/2>$	190300
	-0.2981 -1/2>+0.9530 +7/2>	0.075

The spin Hamiltonian of the Γ_8 level in a cubic field as outlined by Bleaney⁹⁾ applies here. The spectrum has been investigated and corresponds approximately to this Hamiltonian.

$$\mathcal{H} = g\beta(H_iS_i) + p\beta(H_iS_i^3)$$
 $i=x, y, z$.

The observed and calculated g factors are given in Table I¹⁰.

Table I. Measured and calculated g factor of Γ_8 ground state of Nd³⁺.

g exp.	g calc.		
2.26	2.24		
1.10	1.15		
1.45	1.46		
1.6	1.7		
	g exp. 2.26 1.10 1.45 1.6		

(b) The axial field spectrum has been investigated by Bleaney *et al.*³⁾ and remeasured by us¹⁰⁾. The g factors are given $g_{\parallel} = 4.412$ and $g_{\perp} = 1.301$. The ground state can be described possibly by

 $\varphi = a |\pm 3/2 > + b |\pm 1/2 > + c |\mp 7/2 >$

(c) A weak spectrum is also observed with the axial field along the body diagonals. We have, in addition, observed the cubic field spectrum of Ce^{3+} , Dy^{3+} , Tm^{3+} and Gd^{3+} , the axial field spectra of these ions. The axial spectra of ions with even number of electrons such as Tb^{3+} , and Tm^{3+} has been observed as well⁷.

A large part of this work has been performed in collaboration with Mr. M. Dvir, U. Rosenberger and Dr. G. Vincow.

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DISCUSSION

W. P. Wolf: I should like to ask Professor Low what assumption he made with regard to the ratio of fourth to sixth degree crystal fields in deriving the calculated g value for Nd in the cubic site. From the preprint it would appear that you have neglected the admixtures between the two types of terms. When a particular irreducible representation occurs more than once the g values depend explicitly on the ratio B_4/B_5 and the linear interpolation formula given for the energy is then not correct.

W. Low: In the case of Nd³⁺ in CaF₂ it is found that the *g* factor can be fitted well using effectively only the fourth order potential. In Dy³⁺ and Er³⁺ the *g* values of Γ_8 states are very sensitive to the ratio of the fourth to sixth order potential terms. This is generally true for all ions having more than one Γ_8 level. The Nd³⁺ *g* factor is not very sensitive to this ratio.

W.P. Wolf: I cannot agree with your statement about the insensitivity of the g's to the ratio B_4/B_6 even in the case of Nd. For certain values of the ratio some of the g's even change sign and I would suggest that the close agreement with experiment is quite fortuitous.

J. H. Van Vleck: Can you establish at least a qualitative correlation between any of the crystalline fields at the rare earth sites in garnets?

W. Low: We have not yet made the investigation of the correlation between our data on CaF_2 and the rare earth garnets. We feel that we have to understand the spectra and the crystal field in calcium fluoride, which has a simple structure, first before attempting to understand the spectra in the more complicated structure of garnets.

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